

Insertion of Acetylenes into Carbon–Titanium σ -Bonds

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Summary Photochemically induced reactions between $[(\eta^5\text{-C}_5\text{H}_5)_2\text{TiMe}_2]$ and either diphenylacetylene or bis(pentafluorophenyl)acetylene have been shown to produce σ -methylstilbenyltitanium derivatives resulting from insertions of these alkynes into the methyltitanium σ -bond.

In recent years, there has been considerable interest in the so-called insertion reactions^{1†} of unsaturated molecules into metal–carbon σ -bonds of the early transition metals, especially of the Group 4B metals Ti, Zr, and Hf.² Studies of this type may provide a better understanding of the mechanism of Ziegler–Natta catalysis, since this process presumably occurs *via* insertion of a π -co-ordinated olefin into a carbon–titanium σ -bond.³ We now report the first definitive example of the insertion of unsaturated hydrocarbons of the type $\text{RC}\equiv\text{CR}$ into carbon–titanium σ -bonds.

We have previously reported that photochemically induced reactions between $[(\eta^5\text{-C}_5\text{H}_5)_2\text{MMe}_2]$ ($\text{M} = \text{Ti, Zr, Hf}$) and diphenylacetylene in hydrocarbon solvents produce moderate yields of the corresponding metallocycles (Ia–c).⁴ These products, whose structures have recently been conclusively established by X-ray crystallography,⁵ are probably produced by a reaction mechanism involving the transient formation of titanocene, $[(\eta^5\text{-C}_5\text{H}_5)_2\text{Ti}]$, or at least some acetylene-co-ordinated analogue. We now find that such reactions involving $[(\eta^5\text{-C}_5\text{H}_5)_2\text{TiMe}_2]$ and diphenylacetylene can likewise produce a second type of organometallic product (IIa) derived from overall insertion of the acetylene into a carbon–titanium σ -bond of $[(\eta^5\text{-C}_5\text{H}_5)_2\text{TiMe}_2]$.

Thus, photolysis of a dilute solution of $[(\eta^5\text{-C}_5\text{H}_5)_2\text{TiMe}_2]$ and diphenylacetylene (1:2 molar ratio) in hexane solution for 4 h, followed by chromatography on alumina under argon and elution with degassed hexane–benzene mixtures, produces a pale yellow band containing (IIa) followed immediately by a green band containing the titanacycle (Ia). Repeated and careful chromatography is necessary to

† These processes may also be classified as metal–complex 1:2 addition reactions (R. F. Heck, 'Organometallic Chemistry—A Mechanistic Approach,' Academic Press, New York, 1974, p. 32).

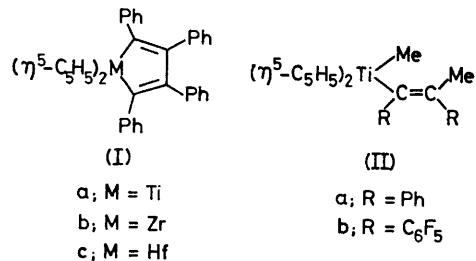
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effect a complete separation of (IIa) (18%) and (Ia) (11%).

The golden-yellow insertion product (IIa) is an exceedingly thermally and oxidatively labile substance, whose structure is based primarily on n.m.r. and mass spectral data [$\tau(\text{C}_6\text{D}_6)$ 10.38 (3H, s, Ti-Me), 8.49 [3H, s, $>\text{C}=\text{C}(\text{Me})-$], 4.06 (10H, s, $\eta^5\text{-C}_5\text{H}_5$), and 2.7-3.3 (10H, m, Ph); m/e 386 (M^+), 371 ($M - \text{Me}$)⁺, 194 $\{(\text{C}_5\text{H}_5)_2\text{TiHMe}\}^+$ and/or $\{\text{PhMeC}=\text{CHPh}\}^+$], as well as on acid hydrolysis to give a mixture (3:1) of *cis*- and *trans*-1,2-diphenylpropene in addition to methane. Since the ¹H n.m.r. spectrum of (IIa) suggests that only a single isomeric organometallic compound is present, it seems likely that the mixture of 1,2-diphenylpropene isomers arises from isomerization of (IIa) during hydrolysis. This suggestion has previously been made for the hydrolysis reaction of the product derived from $[(\text{Ph}_3\text{P})_3\text{RhMe}]$ and diphenylacetylene.⁶ The stereochemical assignment of the insertion product as the *cis*-isomer (IIa) must be regarded as tentative at present, however.

An analogous photochemically induced reaction between $[(\eta^5\text{-C}_5\text{H}_5)_2\text{TiMe}_2]$ and bis(pentafluorophenyl)acetylene, $\text{C}_6\text{F}_5\text{C}\equiv\text{CC}_6\text{F}_5$, has produced a similar insertion product (IIb) [$\tau(\text{CHCl}_3)$ 10.23 (3H, s, Ti-Me), 8.53 [3H, s, $>\text{C}=\text{C}(\text{Me})-$], and 3.64 (10H, s, $\eta^5\text{-C}_5\text{H}_5$); ν (KBr) 2950w (Me), 1525s and 925s (C_6F_5), and 1005s and 810s cm^{-1} (C_6H_5); m/e 566 (M^+), 551 ($M - \text{Me}$)⁺, 374 ($\text{C}_6\text{F}_5\text{CH}=\text{CMeC}_6\text{F}_5$)⁺, 193 $\{(\text{C}_5\text{H}_5)_2\text{TiMe}\}^+$, 178 $\{(\text{C}_5\text{H}_5)_2\text{Ti}\}^+$, and 113 ($\text{C}_5\text{H}_5\text{Ti}\}^+$].[†] As might

be anticipated,⁷ the highly fluorinated product (IIb) exhibits appreciably enhanced stability relative to its hydrocarbon analogue (IIa). It is noteworthy that (IIb) is the sole organometallic product formed during the photolysis of $[(\eta^5\text{-C}_5\text{H}_5)_2\text{TiMe}_2]$ and bis(pentafluorophenyl)acetylene; a titanacycle analogous to (Ia) could not be detected.



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[†] Satisfactory elemental analyses were obtained for this compound.

¹ A. Wojcicki, *Adv. Organometallic Chem.*, 1973, **11**, 88; 1974, **12**, 32.

² P. C. Wailes, H. Weigold, and A. P. Bell, *J. Organometallic Chem.*, 1971, **33**, 181; 1972, **34**, 155; G. Fachinetti and C. Floriani, *ibid.*, 1974, **71**, C5; G. Fachinetti, G. Fochi, and C. Floriani, *J.C.S. Chem. Comm.*, 1976, 230; G. Fachinetti, C. Floriani, F. Marchetti, and S. Merlino, *ibid.*, p. 522; R. J. H. Clarke, J. A. Stockwell, and J. D. Wilkins, *J.C.S. Dalton*, 1976, 120.

³ H. Bestian and K. Clauss, *Angew. Chem. Internat. Edn.*, 1963, **2**, 704; J. Boor, *Ind. and Eng. Chem. (Product Res. and Development)*, 1970, **9**, 437; B. Vollmert, 'Polymer Chemistry,' Springer-Verlag, New York, 1973, p. 193; M. M. Taqui Khan and A. E. Martell, 'Homogeneous Catalysis by Metal Complexes,' Academic Press, New York, 1974, p. 99.

⁴ H. Alt and M. D. Rausch, *J. Amer. Chem. Soc.*, 1974, **96**, 5936.

⁵ J. L. Atwood, W. E. Hunter, H. Alt, and M. D. Rausch, *J. Amer. Chem. Soc.*, 1976, **98**, 2454.

⁶ M. Michman and M. Balog, *J. Organometallic Chem.*, 1971, **31**, 395.

⁷ P. M. Treichel, M. A. Chaudhari, and F. G. A. Stone, *J. Organometallic Chem.*, 1963, **1**, 98; M. D. Rausch, *Inorg. Chem.*, 1963, **3**, 300; R. G. Gastinger, M. D. Rausch, D. A. Sullivan, and G. J. Palenik, *J. Amer. Chem. Soc.*, 1976, **98**, 719; *J. Organometallic Chem.*, 1976, **117**, 355.